## Kinetics of CH<sub>4</sub> oxidation in mixed culture

Yutaka DOTE Department of Civil and Environmental Engineering, Miyazaki University 1-1 Gakuenn Kibanadai Nishi, Miyazaki 889-2155, Japan Tel: +81-985-58-7340, Fax: +81-985+58+7344, e-mail: dote@civil.miyazaki-u.ac.jp

## ABSTRACT

Methane is known as a 'greenhouse' gas and produced in large quantities from a wide range of sources such as swamps, marshes, and paddy fields. Municipal solid waste landfill sites are one of the methane production sources. The main paths of methane emission from a landfill site are cover soils and gas venting pipes. A lot of research on the potential of cover soils to oxidize methane has been reported. Another place where methane can be oxidized is gas-venting pipes. To design a device for methane oxidation in gas venting pipes by aerobic microorganisms, it is necessary to clarify the  $CH_4$  oxidation rate at various  $CH_4$  and  $O_2$ concentrations.

Although many researches on kinetics of  $CH_4$  oxidation were reported, the effect of  $O_2$  concentration was scarcely discussed. The purpose of this paper is to determine the kinetics of  $CH_4$  oxidation and included parameters at various  $CH_4$  and  $O_2$  concentrations.

Cultures used for  $CH_4$  oxidation experiments were prepared by enrichment of extract from a digested sewage sludge, park soil, or leachate from two landfill sites under 20 % of  $CH_4$ , but not isolated. Therefore, the cultures were the mixture of  $CH_4$  oxidizing and other aerobic microorganisms. The  $CH_4$  oxidation experiments were conducted with a batch reactor in which the change in gas volume can be measured. The mixture of air and  $CH_4$  was introduced; initial  $CH_4$  concentrations were ranged from 5 % to 20 %.

CH<sub>4</sub> oxidation was expressed as a Monod equation for CH<sub>4</sub> and first order reaction for O<sub>2</sub>. A determined half saturated constant (K<sub>m</sub>) was  $7.5 \times 10^{-5}$ - $7.1 \times 10^{-4}$  (mol/L), and a maximum oxidation rate (V<sub>max</sub>) was  $3.8 \times 10^{-14}$ - $3.9 \times 10^{-11}$  (mol/hr/cell). The mole ratio of consumed O<sub>2</sub> to CH<sub>4</sub> was 1.6-1.8.